PRODUCT INFORMATION SHEET

Lead-acid battery is not a target product for SDS (safety data sheet). This sheet is intended to be issued in order to provide "reference information" to ensure the safe handling of the product.

1. Chemical Product and Company Identification

Product name	: Lead Acid battery
Information on company	
Company name	: GS Yuasa Battery Ltd.
Relevant dept.	: Sales Planning Group, Sales Planning Division
Address	: 1-7-13, Shiba-Koen, Minato-ku, Tokyo, 105-0011, Japan
Phone number	: +81-3-5402-5733
Fax number	: +81-3-5402-5743

2. Hazards Identification

GHS classification		
Physical and chemical hazards	: Not applicable	
Health hazards	: Not applicable	
Environmental hazards	: Not applicable	
GHS label elements:		
Pictogram	: None	
Signal words	: None	
Hazard statements	: None	
Precautionary statement	: None	
Other risks/hazards	: No information	

3. Composition/Information on Ingredients

l	Information on composition and ingredients:					
	NO.	Chemical name or Component common name part		Content rate (mass ratio)	Chemical formula	CAS no.
	1	1 Lead Terminal, electrode plate			Pb	7439-92-1
	2	Lead dioxide	le Electrode plate		PbO ₂	1309-60-0
	3	Lead sulfate	Electrode plate		PbSO ₄	7446-14-2
	4	Dilute sulfuric acid (27~50%)	Electrolyte	30~45%	H_2SO_4	7664-93-9
	5	Polypropylene or ABS resin	Battery container, lid	4~8%		9003-07-0 /9003-56-9

4. First-aid Measures

If inhaled

: (Lead, lead dioxide, lead sulfate, dilute sulfuric acid)

lf on skin	 Remove person to fresh air, keep comfortable for breathing. Get medical advice/attention. : (Lead, lead dioxide, lead sulfate) Wash skin with plenty of water and soap. If skin irritation occurs, get medical advice/attention. (dilute sulfuric acid) Take off or remove immediately all contaminated clothing. Rinse skin with water or shower. If skin irritation or chemical injury occurs, get medical advice/attention.
If in eyes	 : (Lead, lead dioxide, lead sulfate, dilute sulfuric acid) Open the eyelids with your fingers, rinse thoroughly with water for at least 15 minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Get medical attention/advice.
If swallowed	: (Lead, lead dioxide, lead sulfate) Rinse mouth. Get medical advice/attention. (dilute sulfuric acid) Rinse mouth. Give plenty of water. Do not induce vomiting. Get medical advice/attention.
Most important symptoms/effects, acute and delayed	 : (Lead, lead dioxide, lead sulfate) Stomach cramps, lethargy, headache, nausea, vomiting, weakness, wheezing, pallor, hemoglobinuria, collapse. (dilute sulfuric acid) Corrosive, burning sensation, sore throat, cough, breathlessness, shortness of breath, redness, pain, blisters, severe skin burns, severe burns, abdominal pain, shock or collapse.
Protection for first-aiders	: Rescuers wear protective equipment such as rubber gloves and tight-fitting safety goggles.
Special note to physician	: (Dilute sulfuric acid) Symptoms of lung edema often do not show until a few hours have passed, and it might aggravate if it does not take a rest. Therefore, it is necessary to take a rest and medical observation.

5. Fire Fighting Measures

Suitable extinguishing media	: Extinguish the fire by extinguishers of dry chemical agent,
	foam fire extinguish agent, and non-flammable gas.
Unsuitable extinguishing media	: No information.

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Specific risk/hazard	 In case of fire, there is a possibility that irritative, corrosive or toxic fumes or gases are generated. There is a possibility of explosion of the product by heat.
Specific fire fighting method	 Cut off the power in case of connection/energizing the product into the device, if can be coped with safely. Move the product from the fire area if it is not dangerous. After extinguishing the fire, continue to cool the container thoroughly with plenty of water. Immediately move the movable product to safe place when fire occurs in surrounding. If it is not movable, cool the product with water spray. Keep away the combustible materials to prevent spread fire around.
Protection for fire-fighters	: Extinguish fire from upwind. Wear appropriate protective clothes for chemical (self- contained breathing apparatus, protective glasses, etc.) to fire fighting.

Personal precautions, protective	: Wear appropriate protective equipment (gloves, protective
equipment and emergency measures	glasses, protective clothing and the like), when processing the leakage.
	Do not touch or walk through the leakage.
	Do not breathe dust, mist and vapour.
Precautions for the environment	: Be careful to not discharge the product into the rivers, sewer, and soil.
Method for containment and clean-up	 If dilute sulfuric acid is leaked, stopping the flow with sand and earth, absorbing mat and the like, remove by absorbing with them. And then, neutralized with sodium bicarbonate of slaked lime, and wash off with plenty of water. Absorb by sprinkling misty water when the gas is generated Collected material should be disposed in compliance with '13. Disposal Considerations'.
Prevention of secondary hazards	: Immediately remove all ignition sources in the vicinity. Prepare fire extinguishing equipment just in case it is ignited.

. Handling and Storage	
Handling	
Technical measures	: Take measure described in '8: Exposure Controls and
	Personal Protective Equipment', and wear appropriate protective equipment.
Local exhaust/general ventilation	: Work in a well-ventilated place and provide local exhaust or general ventilation as necessary.
Cautions for Safety Handling	: Do not use fire near the product.

	Do not dismantle or modify the product. Do not do short-circuit between the terminals.
	Handling and charging of the product should be in well ventilated place.
	Prevent falling and overturning of container. Careful to not give a shock.
	Try to not damage the product.
	Be careful that there is a possibility of spewing dilute sulfuric
	acid from the vent plug of product when open.
	Be careful not to spill the dilute sulfuric acid.
	Do not eat, drink or smoke when using this product.
Storage	
Technical measures	: Provide a ventilation and lighting required for storing and
	handling hazardous materials in the storage location.
Storage condition	: Do not store near the fire.
	Do not store in place where is exposed to high temperature,
	high humidity, rain, direct sunlight.
	Store in place where is no risk of fire, toxic gas, liquid
	droplets, generating or invasion of dust, and submerged.

Controlled exposure level	: Lead (electrode plate, terminal), lead dioxide(electrode
	plate), lead sulfate(electrode plate)
	Lead and its compounds(as lead)
	$TLV = 0.05 \text{ mg/m}^3$
Permissible exposure level	
Japan Society for Occupational	: Lead(electrode plate, terminal), lead dioxide(electrode
Health	plate), lead sulfate(electrode plate)
	Lead and its compounds(as lead) : TLV = 0.1 mg/m ³
	Dilute sulfuric acid(electrolyte)
	Sulfuric acid: Maximum permissible exposure limit =
	1mg/m ³
ACGIH	: Lead(electrode plate, terminal), lead dioxide(electrode
	plate),lead sulfate(electrode plate)
	LEAD AND INORGANIC COMPOUNDS, AS Pb
	TLV-TWA = 0.05 mg/m ³
	Dilute sulfuric acid(electrolyte)
	Sulfuric acid: TLV-TWA = 0.2mg/m ³
Engineering controls	: Provide hand wash and eyes wash facilities and safety
	shower near the handling place as necessary.
Personal protective equipment	
Respiratory protection	: Wear respiratory protective equipment (air respirator, dus
	mask, gas mask (for acid gases)) as necessary.
Hand protection	: Wear impermeable protective gloves (acid resistance).
Eye protection	: Wear protective glasses, goggle type safety glasses and

 Skin and body protection
 : Wear protective clothing, protective apron and the like as necessary.

 Hygiene measures
 : Do not eat, drink or smoke when handling.

 Wash hands thoroughly after handling.
 Protective equipment shall be inspected regularly according to the protective equipment checklist.

9. Physical and Chemical properties

Describes the information about the components below.

	Lead	Lead dioxide	Lead sulfate	Dilute sulfuric acid
Appearances	Silver white solid	Brown crystal or	White crystal	Colorless
(physical state, form,		powder		transparent liquid
color, etc.)				
Odor	No information.	No information.	No information.	Odorless (normal
				temperature)
Threshold of odor	No information.	No information.	No information.	No information.
рН	No information.	No information.	No information.	≦1
Melting point	327.4°C	888°C	1170°C	No information
Boiling point, initial	1,749°C	1,480°C	No information	No information
boiling point and				
boiling range				
Flash point	Non flammable	Non flammable	Non flammable	Non flammable
Flammability(solid,	Non flammable	Non flammable	Non flammable	Not applicable
gas)				
Specific	11.35g/cm ³	9.53g/cm ³	6.2	Approx. 1.2~1.4
gravity(density)	(20°C)			
Solubility	Water: Insoluble.	Water: Insoluble.	Water: Hardly	Miscible in water.
			soluble.	Soluble in alcohol.
Partition coefficient (n-	No information.	No information	No information	No information
octanol/water)				
Auto-ignition	Non flammable	Non flammable	Non flammable	Non flammable
temperature				
Decomposition	No information.	290°C	1000°C	No information
temperature				
Viscosity	No information.	No information.	No information.	No information.
Other Information	No information.	No information.	No information.	No information.

10. Stability and Reactivity

Stability

: (lead)

When oxygen is present, it will be eroded by pure water and the weak organic acid. At normal temperature, it will be eroded by fluorine or chlorine.

(lead dioxide/ lead sulfate)

It is considered to be stable under normal handling and

	storage.
	(dilute sulfuric acid)
	At first, vapor is generated by heating, and generate
	sulfuric acid vapors if continue to heat.
	Rapid contact with water might be generate a large amount
	of heat, and sometimes the acid is scattered.
	Dilute sulfuric acid which is generated by diluting with
	water, generates hydrogen gas by the corrosion of various
	metals and may cause flash explosion by mixing with air.
	There is hygroscopic.
Hazardous reactivity	: (lead)
	It does not occur hazardous reaction under normal condition.
	(lead dioxide)
	React violently with combustible materials and organic
	matter (sulfuric acid, hydrogen peroxide, phosphoric acid),
	and it may cause risk of fire.
	(lead sulfate)
	It may react with strong oxidizing agents.
	(dilute sulfuric acid)
	It may cause fire or explosion by many reactions.
	It is strong oxidant and reacts with combustible and
	reducing materials.
	It is strong acid and reacts violently with bases and is
	corrosive to most common metals forming a
	flammable/explosive gas(hydrogen).
	React with water and organic materials violently and
	release heat.
Conditions to avoid	: Heating, contact with ignition sources (open flame, spark,
	etc.,)
Incompatible materials	: (lead): Oxidizing agent.
	(lead dioxide): Flammable materials, reducing materials.
	(lead sulfate): Strong oxidizing agents.
	(dilute sulfuric acid):
	Combustible materials, reducing materials, strong oxidizing agents, strong bases.
Hazardous decomposition products	: In case, there is a possibility that irritative or toxic gases or
	fumes (sulfur trioxide, carbon monoxide, mist sulfate, sulfur
	dioxide, hydrogen sulfide) are generated.
	In case of fire, there is a possibility that irritative or toxic
	gases or fumes are generated.

11. Toxicological Information

Indicate the information for each of components of lead acid battery as below.

\bigcirc Lead (electrode plate, terminal)	
Acute toxicity (Oral)	: No data.
Acute toxicity (Dermal)	: No data.
Acute toxicity (Inhalation: Gases)	: Classification not applicable because it is a solid in the definition of GHS.
Acute toxicity (Inhalation: Vapours)	: No data.
Acute toxicity	: No data.
(Inhalation: Dust and Mists)	
Skin corrosion/irritation	: No data.
Serious eye damage/eye irritation	: No data.
Respiratory or skin sensitization	: No data.
Germ cell mutagenicity	: Although there are contradicting results about the
	chromosome aberration in the peripheral blood lymphocytes
	from people who are engaged in lead-related work (IARC
	suppl.7 (1987), EHC 3 (1977), DFGOTvol.17 (2002), ACGIH
	(7th, 2001)), there are descriptions of lead itself having
	chromosome aberration/micronucleus inductive actions.
	Therefore, it was classified as Category 2.
Carcinogenicity	: From the below classifications, it was classified as Category
	2.
	 IARC Supplement 7 (1987) and Japan Society for
	Occupational Health: 2B
	• ACGIH (7th, 2001): A3
	• EPA (IRIS (1993)): B2
Reproductive toxicity	: Since there is the description that there is the affect for
	sperm formation disorder in human exposure example (EHC
	3 (1977), ACGIH (7th, 2001), DFGOTvol.17 (2002)), and
	ovulation dysfunction was observed in the female
	occupation exposure example (EHC 3 (1977)), it was
	classified as "Category 1A". There are the descriptions
	about the relationship with neonatal developmental disorder
	of cognitive function (ACGIH (7th, 2001), DFGOTvol.17
	(2002), PATTY (4th, 1994) and IARC 23 (1980)), and the
	descriptions about the relationship with the increase of
	miscarriage (DFGOTvol.17 (2002), and PATTY (4th, 1994)).
	However, the distinct conclusion has not obtained.
Specific target organ toxicity	: Although there was a case report that renal dysfunction was
(single exposure)	observed in the acute toxicity in human (DFGOT, vol.17
	(2002)), there was the description that no kidney damage in
	the subsequent epidemiologic study in the same source of
	reference. Therefore, the data is insufficient for considering
	the kidney as target organ, therefore, it is classified as
	"classification not possible".
Specific target organ toxicity	: Due to the descriptions that the target organs were
(repeated exposure)	hematopoietic system, nervous system, kidney, and

cardiovascular system in DFGOTvol.17 (2002), that heme synthesis inhibitors, nephropathy and brain diseases were observed in the human exposure examples in EHC 3 (1977), ACGIH (7th, 2001), PATTY (4th, 1994), and IARC 23 (1980), that it affects to the peripheral nerve and function of central nerve system in humans exposure examples in EHC 3 (1977), ACGIH (7th, 2001), PATTY (4th, 1994), that it affects to cardiovascular system, such as high blood pressure in human exposure examples in EHC 3 (1977), ACGIH (7th, 2001), that the immunosupressive effect was observed in human exposure examples in PATTY (4th, 1994), it is considered that the target organs were hematopoietic system, the kidney, central nervous systems, peripheral nervous system, cardiovascular system and immune system, and they all were classified as "Category 1". Although there are the descriptions of the case reports of thyroid or adrenal hypofunctions in EHC 3 (1977), each case reports are before 1970, and there is no similar report after that, since there is the description that no effects was observed in the thyroid in DFGOTvol.17 (2002), the thyroid and the adrenal gland were not considered as for target organs.

: No data.

: No information.

Lead dioxide (electrode plate)
 Acute toxicity (Oral)
 Acute toxicity (Dermal)
 Acute toxicity (Inhalation: Gases)

Aspiration hazard

Others

Acute toxicity (Inhalation: Vapours) Acute toxicity (Inhalation: Dust and Mists) Skin corrosion/irritation

Serious eye damage/eye irritation

Respiratory or skin sensitization Germ cell mutagenicity

- : No data.
- : No data.
- : Classification not applicable because it is a solid in the definition of GHS.
- : No data.
- : No data.
- : Since there is the description of "Probably a severe eye, skin, and mucous membrane irritant "(HSDB (2006)), it is considered that indicate severe irritation to skin. Therefore, it was classified as Category 2.
- : Since there is the description of "Probably a severe eye, skin, and mucous membrane irritant "(HSDB (2006)), it is considered that indicate severe irritation to eyes. Therefore, it is classified as Category 2A.
- : No data.
- : From the description of NTP DB (Access on February 2006);
 - Heritable germ cell mutagenicity tests: None.
 - Germ cell/ somatic cells in vivo mutagenicity tests:

None.

- Germ cell/ somatic cells in vivo genetic toxicity test: None.
- Positive (strong) results of multiple indicators in in vitro mutagenicity test: None.

It was classified as "Classification not possible".

: From the below classifications, it was classified as Category 2.

- NTP (2005): R
- IARC (1987): Group 2B
- ACGIH (2001): A3
- The Japan Society for Occupational Health : 2B
- : Since lead is known as neurotoxic substance and reproductive toxic substance for human, it is classified as "Category 1A" based on experts' judgement.
- : For this substance, it is assumed that the classification based on the effects of inorganic lead compounds. As the toxicity of inorganic lead compounds for humans, there is the description below;

"acute effects and chronic effects of inorganic lead has been recognized almost the same symptoms. By inhalation or ingestion of inorganic lead, it has been reported that cause the convergence of the mouth, thirst. And also nausea, vomiting, upper abdominal discomfort, loss of appetite, abdominal pain, constipation and the like has been reported as effects on the digestive organs.

Effects on hematopoiesis are typical effects of inorganic lead, it has been observed hemoglobin synthesis inhibition and anemia due to shortened of red blood cell life, caused by inhibition of δ -aminolevulinic acid and heme synthesis enzyme. Interstitial nephropathy as the effect to the kidneys, in addition to decreasing amount of urine, proteinuria, hematuria, urine cylinder, the proximal tubule disorder exhibiting a Fanconi syndrome typified by diabetes and amino acid urine is reported. Inorganic lead affects on the peripheral nervous system, in particular, muscle weakness in limbs, pain and convulsions are observed. In addition, although it is very rare case in adults, in case of being exposed to extremely high concentrations (details unknown), the effects on the central nervous system are observed such as ataxia, headache, paresthesia, depression and coma.

However, in effects on the central nervous system, particularly sensitive in children, and the symptoms with no restless, aggressive personality, difficulty concentrating, decline of memory and the like have become a problem in

Carcinogenicity

Reproductive toxicity

Specific target organ toxicity (single exposure, repeated exposure)

the U.S." (CERI Hazard Assessment Report 2001-9(2002)) Therefore, blood system, kidney and nervous system are considered to be target organs. From the above, it was classified as "Category 1(blood system, kidney, nervous system)". Aspiration hazard : No data. Others : No information. ○Lead sulfate (electrode plates) Acute toxicity (Oral) : No data. Acute toxicity (Dermal) : No data. : Classification not applicable because it is a solid in the Acute toxicity (Inhalation: Gases) definition of GHS. Acute toxicity (Inhalation: Vapours) : No data. Acute toxicity : No data. (Inhalation: Dust and Mists) Skin corrosion/irritation : No data. As effects on humans, although there is no data that can be obtained for local effects on the skin and mucous membranes by lead and inorganic lead compounds, there is the description of that there is likely to cause severe irritation and burns to the skin. Serious eye damage/eye irritation : No data. As effects on humans, although there is no data indicating the local effects on the mucous membranes by lead and inorganic lead compounds, there is the description of that there is likely to cause severe irritation and burns to the eye. Respiratory or skin sensitization : No data. Germ cell mutagenicity : It is "Classification not possible" due to insufficient data of in vivo test. In addition, in the in vitro test, there is the negative report in Ames test. Moreover, although it may not be necessarily matched results have been obtained, there is also the positive result in chromosome analysis using peripheral blood of workers who received the occupational exposure of lead. However, the used method is insufficient in most of the tests, overall it stated that it cannot be conclusive evaluation of genotoxicity in human. Inorganic lead compounds in MAK / BAT (2010) are classified as germ cell mutagenicity 3A. Carcinogenicity : In carcinogenicity evaluation of IARC, it is classified as Group 2A as inorganic compounds. Therefore, this substance was classified as Category 1B. In addition, it is classified as 2B as a lead compound in The Japan Society

for Occupational Health, and A3 as inorganic lead

compound in ACGIH.

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Reproductive toxicity

: Although there is no data of this substance, as effect on humans of inorganic lead compounds, an increase in the spontaneous abortion of pregnancy before 20 weeks was observed by a high concentration exposure of mothers in cases or epidemiological studies.

There is description that exposure during the pregnancy is related increased in teratogenicity, low weight newborns and suppression of body weight gain after birth.

Although decrease in sperm count and semen volume, morphological changes of sperm, and decreasing of sperm motility were observed by the occupational exposure, in most studies, it is observed dose-response relationship between these effects and exposure concentration of lead, and toxicity for sperm has been reported to have been unclear.

Specific target organ toxicity : Although there is no data of this substance, neurotoxic (single exposure) effect of lead is known, and receives the influence of the lead in both the peripheral nerves and the central nervous system. Lead encephalopathy is one of the early symptoms of acute exposure. From also that there is a report of the suppression of the pituitary hands and nerve conduction velocity by the occupational exposure, it is classified as "Category 1 (nervous system)". In high concentration acute exposure of lead and inorganic lead compounds, it causes dysfunction of the proximal tubule. There is the description that it causes Fanconi syndrome (diabetes, amino acid urine disease, phosphate urine disease and the like) as renal symptoms of acute lead poisoning. Therefore, it was classified as "Category 1 (kidney)".

Moreover, lead is also known to give a change in the blood system. There is the description that the hemoglobin synthesis inhibition and small blood cell anemia and hypochromic anemia due to shortened of red blood cell life are caused by δ -aminolevulinic acid and heme synthesis enzyme are inhibited. Therefore, it is classified as "Category 1 (blood system)".

Other, colic is the initial symptoms of occupational exposure or high concentration acute exposure. Since there is the description of the associated symptoms such as constipation, severe abdominal pain, nausea, vomiting, loss of appetite, it is classified as "Category 1 (digestive system)".

: Although there is no data of the substance, there is the description that in high concentration repeated exposure by lead and inorganic lead compounds, it leads to irreversible changes to kidney including tubular atrophy, interstitial

Specific target organ toxicity (repeated exposure)

fibrosis glomerular sclerosis, and eventually it cause chronic nephritis. Therefore, it was classified as "Category 1(kidney)".

In addition, there is the report of the lead epidemiological study, hemoglobin concentration and hematocrit value of poisoning patients were significantly decreased compared with control subjects of non-exposure.

There is the description that the hemoglobin synthesis inhibition and small blood cell anemia and hypochromic anemia due to shortened of red blood cell life are caused by δ -aminolevulinic acid and heme synthesis enzyme are inhibited. Therefore, it was classified as "Category 1 (blood system)".

On the other hand, there is the research study to support the relationship between the chronic lead poisoning and myocardial injury, and there is reported that abnormal electrocardiogram in workers of lead poisoning was observed. In addition, from the data of epidemiological studies, since it has been concluded that internal absorption of lead causes a significant increase in blood pressure in both diastolic and systolic of the heart, it was classified as Category 1(cardiovascular).

Moreover, suppression of motor nerve conduction velocity was observed in worker who blood concentration of lead is high, and also there is a report of the Parkinson's syndrome has been observed in seven out of nine that have been exposed for more than 30 years in a lead-acid battery. Therefore, it was classified as "Category 1(nervous system)".

: No data.

Aspiration hazard

ODilute sulfuric acid (electrolyte)

Others

: No information.

Acute toxicity (Oral)	: Based on the Rat LD ₅₀ value: 2140mg/kg and the description of the death case report by the oral ingestion in humans (amount of intake is unknown), it was classified as Category 5 ("Not classified" by JIS classification).
Acute toxicity (Dermal)	: No data.
Acute toxicity (Inhalation: Gases)	: Classification not applicable because it is a liquid in the definition of GHS.
Acute toxicity (Inhalation: Vapours)	: No data.
Acute toxicity (Inhalation: Dust and Mists)	: Based on rat LC ₅₀ value: 0.375mg/L (4 hour exposure) and 347ppm (1-hour exposure) (4 hour equivalent value: 0.347mg/L), it was classified as Category 2.
Skin corrosion/irritation	: Since pH of concentrated sulfuric acid was 1 or less, it was

	judged to be corrosive substance in accordance with the GHS classification standards, and classified as Category 1A-1C.
Serious eye damage/eye irritation	: There is the description that the critical damage to the eye accompanied by lysis of anterior chamber of eye was observed in accident case of human. And also from the description that the moderate irritation with 5% solution and the severe irritation with 10% solution were observed to the eye of rabbit, therefore, it was classified as "Category 1".
Respiratory or skin sensitization	 Respiratory sensitization: No data. Skin sensitization: There is no test data on skin sensitizing of sulfuric acids. Although sulfuric acid has been industrially used for several decades, there is no case report of skin sensitization while skin injuries by skin irritation are well known. Although an extensive amount of sulfate ion exists internally (the sulfate ion in serum ~33 mmol/L, and 50 times more in cells), allergic reactions do not occur. In allergic test of sulfuric acid salt of metal, even if allergic positive with metal may occur, sulfuric ion is presumed to result in allergic negative as is suggested by the negative results in sulfate of zinc. Based on the description that conclusion is obtained from the results mentioned above that sulfate does not cause allergy to human, it is classified as "Not classified".
Germ cell mutagenicity	: For in vivo, there is not any test data which the reproductive cells and the somatic cells were used. For in vitro mutagenicity tests, there is the positive result only in the test system with the single indicator (chromosomal aberration test). However, there are negative results in other indices. Therefore, it was classified as "Classification not possible".
Carcinogenicity	 Occupational exposure of the mist of the inorganic strong acid including sulfuric acids is classified as group 1 according to IARC, as A2 according to ACGIH, and as K according to NTP. Respect the evaluation of IARC and the latest NTP, it was classified as category 1. However, sulfuric acids itself was classified as the category 4 according to DFGOT. And, since none of those institutions have carried out the carcinogenic classification, it was classified as "Classification not passible"
Reproductive toxicity	possible". : In inhalation exposure test using rabbit and mouse in fetal organogenesis period, it is not observed of fetotoxicity and teratogenicity at the dose causing no maternal toxicity in both species. And also, the effect on the reproductive organ of both sexes is not observed in chronic toxicity test and

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Specific target organ toxicity (single exposure)	 carcinogenicity test. Since the direct effect by irritation/corrosive is the main toxicity, it is judged that there is no concern that indicates the reproductive toxicity, therefore, it was classified as "Not classified". There is the descriptions that in the inhalation exposure of low concentration in humans, airway irritation symptoms such as cough and breath shortness are observed and at high concentration exposure, addition to acute effects such as cough, breath shortness and hemoptysis shedding etc., permanent effects such as functional depression of lungs, fibrosis and emphysema are observed. Additionally, there is the description that hemorrhage and dysfunction in lungs were observed in 8-hour inhalation exposure using guinea pigs. Based on these descriptions, it was classified as "Category 1 (respiratory systems)".
Specific target organ toxicity (repeated exposure)	 In the 28-day inhalation exposure test using rat, cell proliferation in laryngeal mucosa is observed in guidance value range of Category 1. In the 14 to 139-day repeated inhalation exposure test using the guinea pigs, respiratory and lung disorder, such as nasal-septum dropsy, pulmonary emphysema, atelectasis, hyperemia, dropsy, bleeding and thrombosis of bronchioles are observed at the concentration range of guidance value of Category 1. Furthermore, in the 78-week inhalation exposure test using a cynomolgus, histological change as hyperplasia of a cell, the wall thickening, etc. in bronchioles of lungs were observed at the dosage (0.048 mg/L, 23.5 Hr/Day) of the guidance value range of Category 1. From the above, it was classified as Category 1 (respiratory systems). No data.
Others	: No information.

12. Ecological Information

Indicate the information for each of components of lead acid battery as below.

\bigcirc Lead (electrode plate, terminal)	
Ecotoxicity	: No data.
Persistence/degradability	: No data.
Bioaccumulation	: No data.
Mobility in soil	: No information.
Hazardous to the ozone layer	: Not contain ingredients listed in the Annex of the Montreal
	Protocol.
OLead dioxide (electrode plate)	
Ecotoxicity	: No data.

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Persistence/degradability	: No data.
Bioaccumulation	: No data.
Mobility in soil	: No information.
Hazardous to the ozone layer	: Not contain ingredients listed in the Annex of the Montreal
	Protocol.
⊖Lead sulfate	
Ecotoxicity	: Crustacean: Daphnia magna, 48hr-IC₅₀ = 0.5mg/L
	(Acute hazardous to the aquatic environment : Category 1)
	Reliable chronic toxicity data has not been obtained. Since it
	is metal compound, the behavior in water is not known.
	Because acute toxicity is category 1, chronic hazardous to
	the aquatic environment was classified as "Category 1".
Persistence/degradability	: No data.
Bioaccumulation	: No data.
Mobility in soil	: No information.
Hazardous to the ozone layer	: Not contain ingredients listed in the Annex of the Montreal
	Protocol.
⊖Dilute sulfuric acid (electrolyte)	
Ecotoxicity	: Fishes: Bluegill, 96hr-LC ₅₀ = 16-28mg/L
	(Acute hazardous to the aquatic environment : Category 3)
	Toxicity factor is considered to be aqueous solution which
	becomes strong acid, but toxic effect is eased by the buffer
	action in the environmental water. Therefore, Chronic
	hazardous to the aquatic environment was classified as "Not classified".
Persistence/degradability	: No data.
Bioaccumulation	: No data.
Mobility in soil	: No data.
Hazardous to the ozone layer	: Not contain ingredients listed in the Annex of the Montreal Protocol.

13. Precautions for Disposal

•	
Disposal considerations	: In the disposal, follow "Waste Management and Public
	Cleansing Law" and the standards of the local government.
	Entrust disposal to industrial waste disposal contractor who
	have obtained a license from local governor, otherwise if the
	local government is performing waste disposal, entrust them
	disposal.

14. Transport Information

International regulations(dangerous goo	ds)
Inland transport	: Follow the regulation under ADR/RID.
Sea transport	: Follow the regulation under IMO.

Air transport	: Follow the regulation under ICAO/IATA.				
UN number	: 2794				
UN class	: Corrosive substance/Class 8				
Proper shipping name	: BATTERIES, WET, FILLED WITH ACID, electric storage				
Packing group	:-				
Special requirements	: IMO SP295				
	IATA A51, A164, A183, A802				
Marine pollutant	: Not applicable				
Japan domestic regulations					
Inland transport	: Follow the regulation of Fire Service Act and Poisonous and				
	Deleterious Substances Control Law.				
Sea transport	: Follow the regulation of Ship Safety Law.				
Air transport	: Follow the regulation of Civil Aeronautics Act.				
Special safety measures and condition	: Avoid mixed load with other substances as much as				
for transport	possible.				
Handle the dilute sulfuric acid so as not to leak by					
	overturning or falling.				
	Load to not overturning, falling and damage, and take				
	prevention of cargo collapse securely.				
	Avoid transport under the direct sunlight and high				
	temperature.				
	Transport in accordance with the standards of other related				
	laws and regulations.				
Emergency response guideline number(North America)	: 154				

15. Regulatory Information

There are not laws and regulations applicable for the lead-acid battery itself. Describe the information about the components below.

Industrial Safety and Health Act : Dangerous and Harmful Substances Subject to Indicate Their Names (Article 57 of the Act) Lead components: lead dioxide, lead sulfate -Dangerous and Harmful Substances Subject to Notify Their Names (Article 57-2 of the Act) Lead and its inorganic compounds : lead, lead dioxide, lead sulfate Sulfuric acid: dilute sulfuric acid. -Corrosive liquids (Article 326 of the regulation) Sulfuric acid: dilute sulfuric acid Specified Chemical Substances (Article 2 of The Ordinance on Prevention of Dangers Due to Specified Chemical Substances, paragraph 1) Category 3 substance

	- Sulfuric acid: dilute sulfuric acid
	Lead (Appended Table 4 of the Enforcement Ordinance No.1 of
	Article 1 of the Ordinance on Prevention of Lead Poisoning.)
	- Lead
	Lead compounds (Appended Table 4 of the Enforcement
	Ordinance No. 4 of Article 1 of the Ordinance on Prevention of
	Lead Poisoning.)
	- Lead dioxide, Lead sulfate
Labor Standards Act	: Illness chemical substances (Article 75 of the Act, paragraph 2,
	article 35 of the Enforcement Regulations, Appended table of 1-
	2, no.4)
	- Lead and its compounds
	Lead, lead dioxide, lead sulfate
Poisonous and Deleterious	: Deleterious Substances (Article 2 of the designated ordinance)
Substances Control Act	- Lead compounds: lead dioxide
	- Sulfuric acid: dilute sulfuric acid
Act on Confirmation, etc. of Release	: Class I Designated Chemical Substances (Paragraph 2 of
Amounts of Specific Chemical	Article 2 of the Act, Appended Table 1 of Article 1 of the
Substances in the Environment and	Enforcement Ordinance)
Promotion of Improvements to the	- Lead
Management Thereof	Specified Class I Designated Chemical Substances (Paragraph
	2 of Article 2 of the Act, Article 4 of the Enforcement Ordinance)
	- Lead compounds: lead dioxide, lead sulfate
Fire Service Act	: Hazardous material Category 1 Oxidizing solid
	- Chromium, lead or oxide of iodine
	- Lead dioxide
	: Substances Inhibiting Fire Fighting
	- Lead dioxide
	- Sulfuric acid: dilute sulfuric acid
Marine Pollution Prevention Law	: Noxious liquid substances
	- Category Y: Dilute sulfuric acid
Civil Aeronautics Act	: Article 194 of Enforcement Regulations, Appended Table 1 of
	the Notification for Establishing Standards for the Carriage of
	Dangerous Goods in Ships
	- Oxidizing substances: lead dioxide
	 Corrosive substance: lead sulfate (which the content of free acid is more than 3 wt%)
	- Corrosive substance: dilute sulfuric acid
Ship Safety Act	: Article 2,3 of the Dangerous Goods Regulations, Appended
	Table 1 of the Notification for Establishing Standards for the
	Carriage of Dangerous Goods in Ships
	- Oxidizing substances: lead dioxide
	 Corrosive substance: lead sulfate (which the content of
	free acid is more than 3 wt%)
	- Corrosive substances: dilute sulfuric acid
Water Pollution Prevention Act	: Noxious substances (No.1 of Paragraph 2 of Article 2 of the

Water Pollution Prevention Act : Noxious substances (No.1 of Paragraph 2 of Article 2 of the

Act)

- Lead and its compounds :
 - Lead, lead dioxide, lead sulfate
- Corrosive substances: dilute sulfuric acid
- : Designated substances (Paragraph 4 of Article 2 of the Act)
 - Sulfuric acid: dilute sulfuric acid
- Oxidizing substances: lead dioxide
 - Corrosive substance: lead sulfate (which the content of free acid is more than 3 wt%)
- Corrosive substances: dilute sulfuric acid

16. Other Information

Act on Port Regulation

Electrochemical reaction formula:							
Positive	Electrolyte N	legative		Positive	Electroly	te	Negative
PbO ₂ +	$2H_2SO_4 \hspace{0.1in} + \hspace{0.1in}$	Pb	Charge<>Discharge	PbSO ₄ +	$2H_2O$	+	PbSO ₄
Lead Dioxide	Sulfuric Acid	Lead		Lead sulfate	Water	L	ead sulfate

Reference:

Globally Harmonized System of classification and labelling of chemicals, (5th ed., 2013), UN JIS Z 7253:2012

- 1) NITE GHS classification data.
- 2) ECHA Home page (http://echa.europa.eu/information-on-chemicals)
- 3) NITE CHRIP (http://www.safe.nite.go.jp/japan/sougou/view/SystemTop_jp.faces)
- 4) Ministry of Health, Labour and Welfare (<u>https://anzeninfo.mhlw.go.jp</u>)

Notice:

The contents described in this SDS are prepared based on the data and information currently available to us. However, it does not intend to be any guarantees in regard to content, physical and chemical properties, hazards, etc.

Please handle this product in the responsibility of the user after referring to this SDS.

In addition, the precautions are intended for normal handling. Please use under implementing safety measures that are suitable for application/usage if you want to special handling.